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We claim:

- 1. A process for producing dimethyl ether comprising: reacting syngas in contact with a hybrid catalyst in a microchannel reactor at a temperature from about 200° C. to about 400° C. with a contact time of from about 25 milliseconds to less than about 1 second; and converting more than about 60% of CO in the syngas to dimethyl ether and methanol.
- 2. The process of claim 1 comprising converting more than about 60% of CO in the syngas to dimethyl ether.
- 3. The process of claim 1, wherein there is no liquid holdup inside the microchannel reactor.
- **4**. The process of claim **1**, wherein the hybrid catalyst comprises a methanol synthesis catalyst and a methanol dehydration catalyst present in a weight ratio of about 1:1.
- **5**. The process of claim **1**, wherein the hybrid catalyst comprises a methanol synthesis catalyst and a zeolite catalyst present in a weight ratio of about 1:1 to about 2:1.
- **6**. The process of claim **1**, wherein the hybrid catalyst comprises F51-8PPT and ZSM-5 zeolite with a Si/Al ratio of about 30.
- 7. The process of claim 6, wherein the F51-8PPT and ZSM-5 zeolite are present in a weight ratio of about 1:1.
- **8**. The process of claim **1**, wherein the process is carried out in multiple microchannel reactors to produce greater than about 70% CO conversion to dimethyl ether and methanol.
- 9. The process of claim 1, wherein the process is carried out in multiple microchannel reactors to produce from about 80% CO conversion to about 88% CO conversion to dimethyl ether and methanol.
- 10. The process of claim 1, wherein the process is carried out at a temperature of from about  $240^{\circ}$  C. to about  $290^{\circ}$  C. with a contact time of less than about 500 milliseconds.
- 11. The process of claim 1, wherein the source of syngas is biomass.
- 12. A method of making DME, comprising: passing syngas in contact with a hybrid catalyst comprising a methanol synthesis catalyst and a dehydration catalyst in a microchannel reactor at a temperature of from about 240° C. to about 290° C. with a contact time of less than about 1 second and at a pressure in the range of about 0.5 MPa to about 8 MPa; and producing greater than about 70% CO conversion to DME and methanol.
- 13. The process of claim 12, wherein the methanol synthesis and dehydration catalyst comprises a methanol synthesis catalyst and a zeolite catalyst that have been mixed together.
- 14. The process of claim 12, wherein the methanol synthesis and dehydration catalyst comprises a copper or palladium based methanol synthesis catalyst and a zeolite catalyst having a Si/Al ratio of about 30.
- 15. The process of claim 12, wherein the methanol synthesis and dehydration catalyst are present in a weight ratio of about 1:1 to about 2:1.

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- **16**. The process of claim **12**, wherein the methanol synthesis and dehydration catalyst comprise a copper-based methanol synthesis catalyst and a zeolite catalyst present in a weight ratio of about 2:1 to about 1:1.
- 17. The process of claim 12, wherein the contact time is less than about 500 milliseconds and the pressure is from about 2 to about 8 MPa.
- 18. The process of claim 12, wherein the source of syngas is biomass.
- 19. A method of producing dimethyl ether, comprising: providing a syngas source;
  - combining the syngas source with a hybrid catalyst comprising a methanol synthesis catalyst and a methanol dehydration catalyst; and
  - reacting the syngas in the presence of the catalyst in a microchannel reactor under substantially isothermal conditions at a temperature of from about 200° C. to about 350° C. with a contact time of from about 25 milliseconds to about 500 milliseconds to produce DME with greater than about 60% CO conversion to dimethyl ether and methanol.
- **20**. A method of producing dimethyl ether, comprising: providing a biomass syngas source;
- combining the biomass syngas source with a methanol synthesis catalyst and dehydration catalyst, in a microchannel reactor, each microchannel at a single temperature of from about 200° C. to about 350° C. and with a contact time of less than about 1 second; and
- producing greater than about 60% CO conversion to dimethyl ether and methanol.
- 21. A method of producing dimethyl ether, comprising: providing a biomass syngas source;
- combining the biomass syngas source with a methanol synthesis catalyst and zeolite catalyst present in a weight ratio of about 1:1; and reacting the biomass syngas in the presence of the catalyst in a microchannel reactor, at a temperature of from about 240° C. to about 290° C., the temperature varying less than about 2° C. while the biomass syngas is reacting in the microchannel reactor, and with a contact time of less than about 500 milliseconds; and

producing greater than about 60% CO conversion to dimethyl ether and methanol.

- 22. The process of claim 1 further comprising:
- providing the syngas and the bydrid catalyst to a module including a plurality of microchannel reactors; and
- reacting the syngas and the hybrid catalyst in the microchannel reactors at temperatures of from about 200° C. to about 350° C. and with a contact time of less than about 1 second to obtain greater than about 70% CO conversion to dimethyl ether and methanol.

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